SYNTHESIS OF 2,3- AND 2,7-DIHYDROTHIEPINES

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One of the methods used in carbocyclic chemistry to achieve ring enlargement consists in the cycloaddition of an activated acetylene such as (di)methyl acetylene-(di)carboxylate to an enamine of a cyclic ketone¹. In a subsequent step the cyclobutene ring is opened simply by heating the bicyclic intermediate⁺.

We have applied this method to the synthesis of dihydrothiepines, a novel class of compounds⁺⁺. The cycloaddition of enamines of (alkylated) tetrahydrothiophen-3-ones (I and II)⁴ to dimethyl acetylenedicarboxylate in diethyl ether as a solvent proceeded smoothly at room temperature and yielded crystalline 2-(or 3-)thiabicyclo [3.2.0] hept-6-ene derivatives (III and IV). Upon being heated at 100 °C in dioxane these products gave the corresponding dihydrothiepines (V and VI) in an overall yield of 40-90%.



(i) 1 eq. dimethyl acetylenedicarboxylate in diethyl ether, 20 °C; 16 h

(ii) dioxane; 100 °C; 2 h

The compounds were identified by the NMR signals of their vinyl protons: compounds of type (V) gave a singlet in the δ 7.0-7.6 ppm range and those of type (VI) a triplet in the δ 6.6-7.0 ppm range (J ~ 8 Hz).

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A further investigation of the general characteristics of the reaction revealed the following: (a) In the cycloaddition the enamine double bond is more reactive in the 4,5- than in the

3,4-position. This was concluded from the reactions with enamines of tetrahydrothiophen-3-one (mixtures of (VII) and (VIII), ratio ~ 2 : 1). A mixture of 2,7- and 2,3-dihydrothiepines (IX and X) was obtained in a ratio ~ 1 : 2. Since the enamines are not interconvertible under the conditions applied in this experiment, this result indicates that the 2,3-dihydro compound (VIII) is about four times as reactive as its 2,5-dihydro isomer (VII).



- (b) Alkyl groups adjacent to the enamine double bond lower the rate of cycloaddition.
- (c) As expected, the effect of the secondary amino group on the rate of cycloaddition decreases in the order: pyrrolidino > morpholino ≥ piperidino.
- (d) Dimethyl acetylenedicarboxylate is more reactive than methyl propiolate, which reacts only with pyrrolidine enamines.

Acid-catalysed hydrolysis of the enamine moiety gave enols stabilized by hydrogen bonding, and/or the corresponding ketones. These compounds can be oxidized by m-chloroperbenzoic acid at room temperature to the corresponding sulphones.

REFERENCES AND FOOTNOTES

A.G. Cook, "Enamines: Synthesis, Structure and Reactions", Marcel Dekker, New York, 1969, 230.
M.S. Lin and V. Snieckus, <u>J. Org. Chem.</u> <u>36</u> 645 (1971).

3. J.M. Hoffman and R.H. Schlessinger, J. Am. Chem. Soc. <u>92</u> 5263 (1970).

4. F.A. Buiter, J.H. Sperna Weiland and H. Wijnberg, Rec. Trav. Chim. 83 1160 (1964).

- + Recently, similar reactions have been used for the synthesis of benzazepines (ref. 2).
- ++ Monocyclic thiepines and their dihydro derivatives have not been described; owing to their anti-aromatic character, thiepines are expected to be reactive species (ref. 3).